

Results of UREX Hot Demonstration



Tracy Rudisill

Westinghouse Savannah River Company

January 22, 2003

Advanced Fuel Cycle Initiative
Semiannual Review Meeting
Albuquerque, NM

Background

- ◆ Transmutation of Waste is being Developed by the Advanced Fuel Cycle Initiative
 - Address Disposal of Commercial Nuclear Fuel
 - Improve Performance of the Geologic Repository
- ◆ Transmutation Program will Require Separation of Reactor Fuel into . . .
 - TRU Product Stream which is Converted to Fuel and Transmuted by Fissioning to Generate Power
 - Separate ^{99}Tc and ^{129}I Streams which are Converted to Targets and Transmuted to Short-lived Nuclides
 - Uranium Product that Meets Criteria for Class C LLW

UREX Process

- ◆ The PUREX Process is a Mature Solvent Extraction Process for Irradiated Fuel Designed to Recover Pu and U
- ◆ A Variation of the PUREX Process was Conceived to Treat Large Quantities of Irradiated Spent Fuel and to Provide High Selectivity
 - PUREX was Modified so Only U and Tc are Extracted and the TRU Isotopes (Np, Pu, Am, and Cm) are Rejected with the Fission Products
 - This Uranium Extraction Process is Called UREX

UREX Process Goals

◆ Recovery Efficiencies

- Recover >99.9% of the U and >95% of the Tc
- Reject >99.9% of the TRU Isotopes to the Acid Waste (Raffinate)
- U Product Meets Requirements for Class C LLW
 - Fission Products (Primarily Cs and Sr)
 - TRU Isotopes (<100 nCi/g)

◆ Waste Minimization

- Conversion of All Chemicals to Gases During Subsequent Processing
- Acetohydroxamic Acid (AHA) Used to Prevent the Extraction of Pu and Np

Chemistry of AHA in the UREX Process

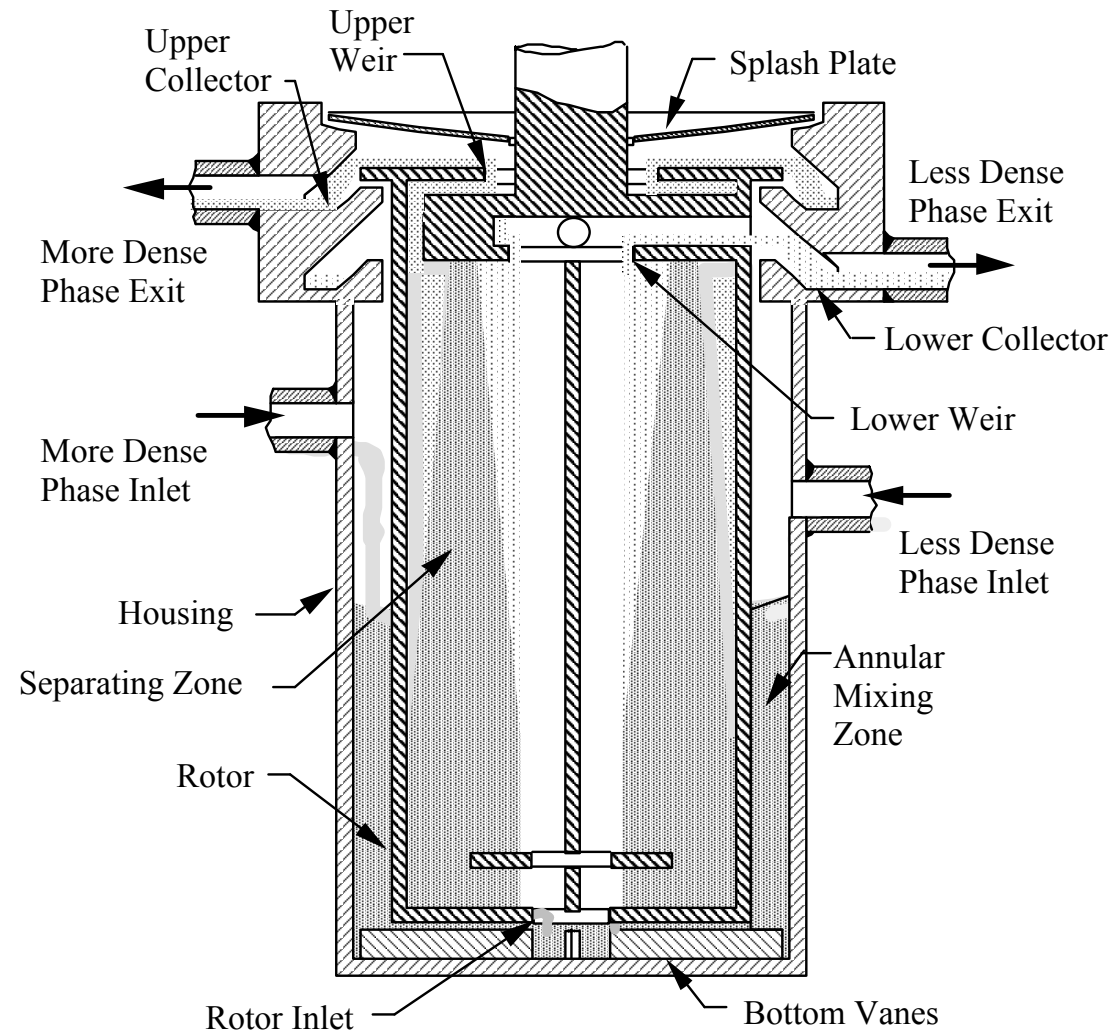
- ◆ AHA is an Analogue of Hydroxylamine
 - Hydroxylamine Nitrate is used as a Reductant in Nuclear Processing
- ◆ AHA Serves as Both a Reductant and a Complexant in the UREX Process
 - Reduces Np (VI) to Inextractable Np (V)
 - Prevents Extraction by Complexing Pu (IV) and Np (IV)
- ◆ AHA Readily Decomposes to Gaseous Products During Waste Evaporation
- ◆ AHA Hydrolyzes in Acidic Solutions

Hot UREX Demonstration

- ◆ A “Hot” Demonstration of the UREX Process was Performed in the SRTC Shielded Cells Facility Using Spent Fuel from the Dresden Reactor
- ◆ UREX Demonstration Performed with Centrifugal Contactors Previously Used to Demonstrate the Extraction of Cs from SRS Waste

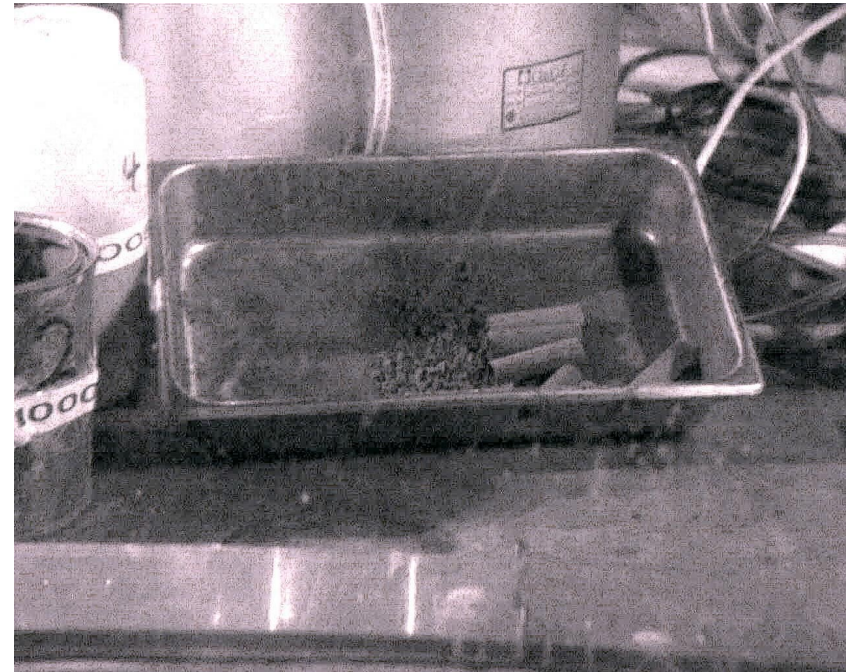


Centrifugal Contactor Schematic



Dresden Reactor Fuel

- ◆ Boiling-Water Reactor
Near Morris, IL
- ◆ Fuel Discharged on
September 1, 1975
- ◆ Fuel Burnup 23,480
MWD/MT
- ◆ Fuel Contained 4 kg of
U and 29 g of Pu



Fuel Dissolution

- ◆ Fuel Segments were Dissolved using a 6 L Glass Dissolver and Heating Mantle
 - Stainless Steel Basket used to Contain the Fuel
 - Offgas was Condensed with Water
- ◆ Two Step Dissolution Performed using HNO_3 at 90°C
 - Initial Dissolution Performed in 1-4M HNO_3
 - Following Temperature Spike 10M HNO_3 was Added to Dissolver to Complete Dissolution
- ◆ Hulls were Leached with 4M HNO_3 at 90°C for 2-6 h



Flow Rate Control for Hot UREX Demonstration

- ◆ Positive Displacement and Peristaltic Feed Pumps were Manually Controlled
- ◆ 2 or 4 L Graduated Cylinders with Attached Burette used for Flow Measurements
- ◆ Goal - Measure Flow Rates Once per Hour or As Needed



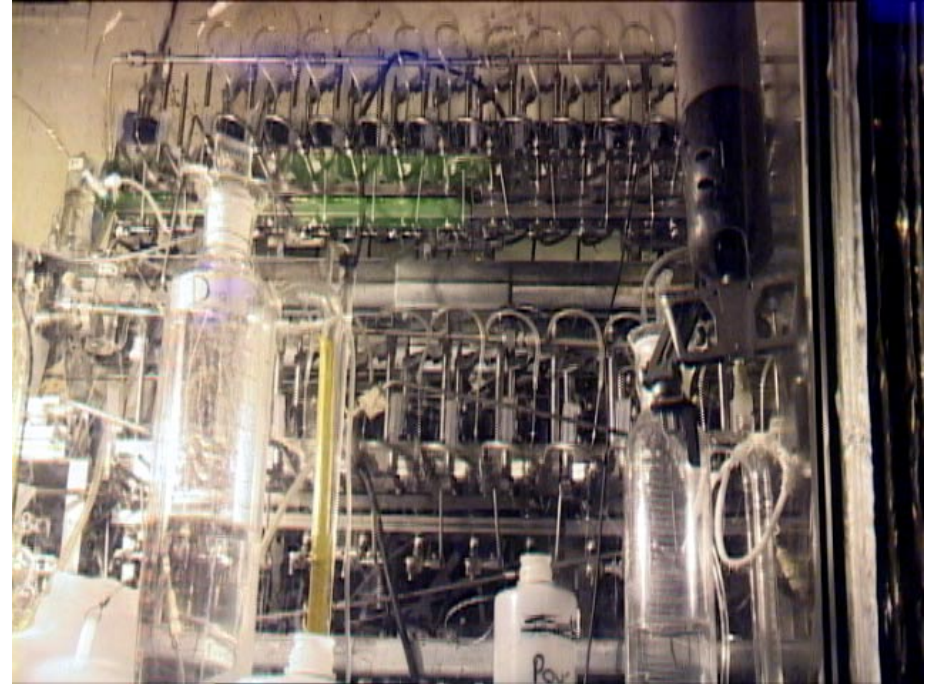
Hot Demonstration Tests

- ◆ 3 Hot Tests Performed

- Two 6 h Tests
- One 30 h Test with Changes in Flow Conditions

- ◆ 30 h Test

- Test 3A Performed at Same Flow Conditions as Tests 1 and 2 but for an 8 h Duration
- Test 3B Performed at Flow Conditions which Generated More Concentrated U and Tc Products

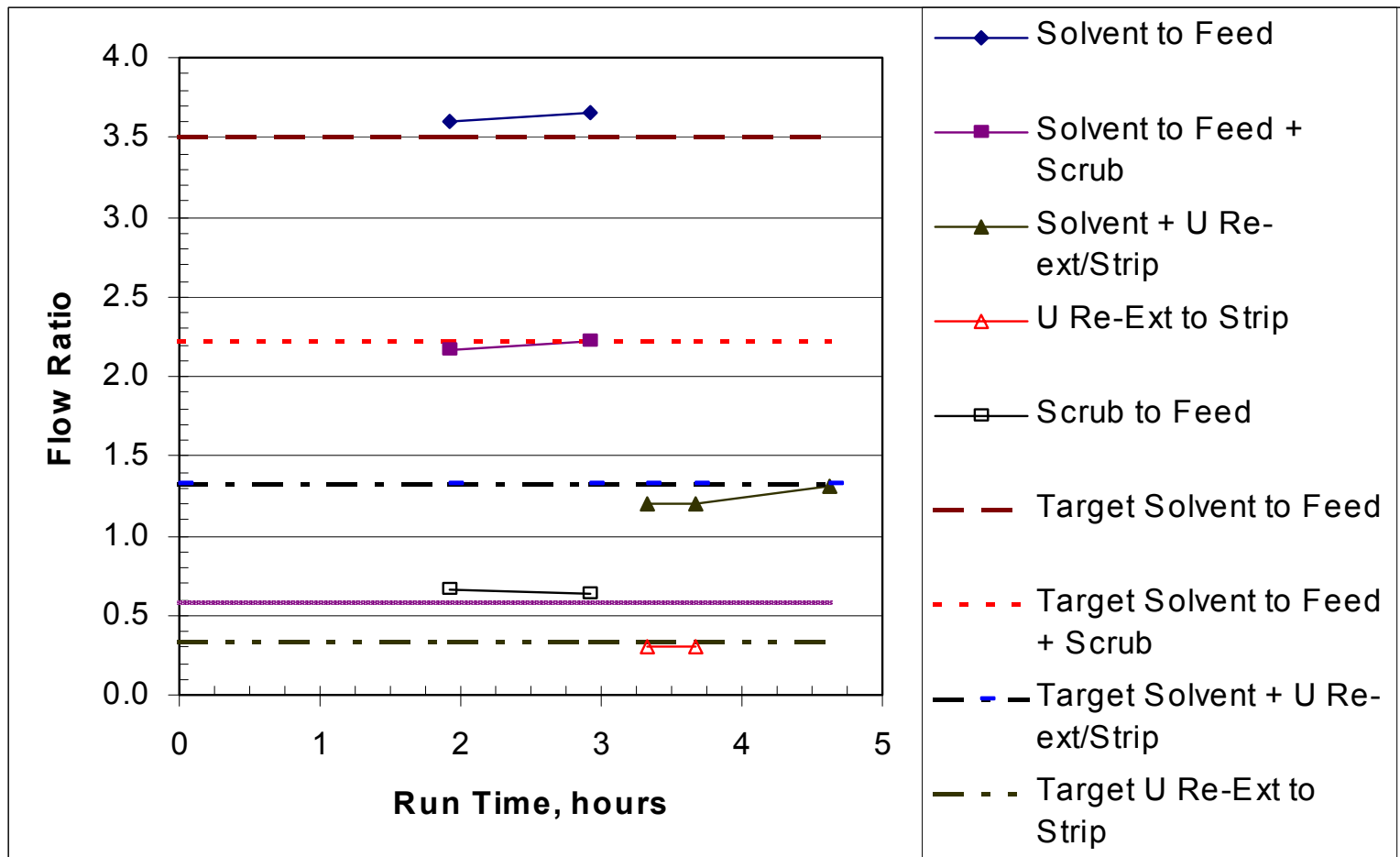


Analytical Methods

Analysis	Method
U	Spectrophotometrically (1-10 g/L U) ICP-MS (Trace Concentrations) Laser Fluorescence (Trace Concentrations)
⁹⁰ Sr	Beta Counting (Ion Exchange Chromatography)
Pu	Alpha Pulse Height Analysis (TTA Extraction)
Tc	Beta Counting (Ion Exchange Separation)
Gamma Emitters	Gamma Pulse Height Analysis
Other Elements	ICP-ES or ICP-MS

Flow Rate Ratios

UREX Test 2



Analytical Results

UREX Test 2

Stream	Time (hr)	U (g/L)	Pu (d/m/mL)	Am (d/m/mL)	Tc (d/m/mL)	Cs (d/m/mL)	Sr (d/m/mL)
U Product	0	3.9	4.12E+04	1.16E+04	2.11E+03	7.90E+04	1.02E+04
	1	51.2	4.43E+03	2.85E+03	2.72E+03	6.05E+03	3.84E+03
	3	54.5	6.01E+03	NA	1.15E+04	8.98E+03	3.85E+03
	4	53.7	1.18E+04	1.53E+04	2.07E+03	1.17E+04	3.85E+03
	6	55.1	3.02E+03	5.07E+03	1.29E+03	1.67E+04	5.00E+03
Tc Product	0	0.0045	NA	NA	2.25E+04	1.53E+05	2.10E+04
	2	0.0045	2.41E+03	NA	1.58E+06	1.28E+05	1.46E+04
	3	0.0045	2.31E+04	1.85E+04	1.82E+06	4.48E+05	2.30E+05
	4	0.0045	8.41E+03	5.00E+03	1.81E+06	3.34E+05	1.02E+05
	6	0.000079	4.67E+03	7.53E+03	1.45E+06	3.93E+05	1.11E+05
Waste Raffinate	1	NA	6.72E+08	8.79E+08	1.64E+05	1.77E+10	NA
	2	NA	8.55E+08	6.57E+08	5.86E+05	1.26E+10	9.78E+9
	3	NA	1.04E+09	NA	5.74E+05	1.93E+10	1.30E+10
	4	0.0026	7.57E+08	NA	1.46E+05	1.68E+10	1.94E+10
	6	0.0297	7.97E+08	7.50E+08	2.91E+05	1.89E+10	1.47E+10

NA - No Analysis

Material Balance Results

UREX Test 2

Stream	Time (hr)	U (%)	Pu (%)	Am (%)	Tc (%)	Cs (%)	Sr (%)
U Product	0	6	0.0187	0.00513	<0.17	0.00181	0.000224
	1	84	0.00201	0.00126	<0.21	0.000138	0.000084
	3	90	0.00276	NA	<0.91	0.000208	0.000085
	4	89	0.00542	0.00682	<0.16	0.000271	0.000085
	6	91	0.00138	0.00226	<0.10	0.000386	0.000111
Tc Product	0	<0.0061	NA	NA	1.4	0.00286	0.000376
	2	<0.0061	0.000895	NA	101	0.00239	0.000243
	3	<0.0061	0.00868	0.00676	118	0.00848	0.00388
	4	<0.0061	0.00314	0.00182	116	0.00629	0.00171
	6	0.000079	0.00161	0.00252	86	0.00682	0.00171
Waste Raffinate	1	NA	103	131	4	136	NA
	2	NA	131	98	15	97	72
	3	NA	157	NA	15	146	94
	4	0.0014	114	NA	4	127	141
	6	0.0161	120	110	8	143	107

NA - No Analysis

Contamination Levels

UREX Uranium Product Stream

Nuclides	LLW Limit, Ci/m ³			UREX Test Results ⁽¹⁾			
	Class A	Class B	Class C	Test 1	Test 2	Test 3A	Test 3B
⁹⁹ Tc	700		3	0.34	0.03	NA	NA
¹²⁹ I			0.08	NM	NM	NM	NM
Nuclides (half-life <5yr)				0.05	ND	ND	ND
⁹⁰ Sr			7000	0.22	0.12	<0.07	<0.07
¹³⁷ Cs	1	44	4600	0.47	0.41	0.04	0.03-0.11
TRU Isotopes (half-life >5yr)			100 ηCi/g	<296	65	16	29-153
²⁴¹ Pu			3500 ηCi/g	1420	214	71	147-1020
²⁴² Cm			20000 ηCi/g	ND	ND	ND	ND

(1) Assumes density of bulk UO₃ is half the crystal density or 3.645 g/cm₃

NA - Not Available

ND - Not Detected

NM - Not Measured

Uranium Stripping

- ◆ Number of Stages in Cell was Insufficient to Strip U from the Solvent
- ◆ 4 Centrifugal Contactor Stages Set-up in Hood
 - U Stripped from Solvent with 0.01M Nitric Acid
- ◆ Whole Body Dose Rate of Bottles Containing U-Loaded Solvent was <1 mRad/hr
 - Low Dose Rate Demonstrates Good Fission Product Decontamination Achieved by UREX Process
- ◆ UO_3 Produced by Evaporation, Precipitation of $(\text{NH}_4)_2\text{U}_2\text{O}_7$, and Calcination



UO_3 Product



Conclusions

- ◆ Demonstration of UREX Process at Baseline Conditions Showed All Goals for Recovery and Decontamination are Achievable
 - U Product was Class C (or Lower) LLW
 - Recovery of >99.9% U and >95% Tc
 - Rejection of >99.9% of Pu to the Raffinate
- ◆ Tc Losses to U Stream were 1.2% in 1st Test and 0.1% in 2nd Test
- ◆ Loss of Pu and Other Actinides to the Tc and U Product Streams was <0.1% in All Tests

Flowsheet for UREX Test 3B

